

# A Short Note on Magnesium Alloys

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## Introduction

This article initially describes the anomalous/problematic properties of magnesium before presenting the latest strategy of stacking fault energy-based alloying element selection to decrease or remove this problem. Understanding the free electron density distribution around atoms in a solid solution is required for stacking fault energy estimations using initial approaches [1]. As a result, the function of atoms has been reconsidered by taking into account the potential of short range order generation rather than a random solid solution. There are two sorts of SROs that have been suggested. In a previously unknown model, the necessary electrical interactions between the host Mg and the alloying element atoms are more fully integrated.

## Discussion

This more successful strategy has also been discussed. An assessment based on these premises has been offered in terms of their successes in mending the problematic aspects of Mg alloys, introducing the comparatively more current Mg alloy systems [2]. The variety of alloy systems studied includes Mg doping, dilute alloy systems, and some rich alloy systems with exceptional features. An unusual addition, doping with oxygen, and its repercussions, has been given in the first category. The potentials of dilute alloy systems and their compositional design based on SRO and SFE have been examined. An unusual addition, doping with oxygen, and its repercussions, has been given in the first category. The potentials of dilute alloy systems and their compositional design based on SRO and SFE have been examined [3]. The most fascinating precipitate systems, involving order and intermetallic formations, long-period stacking order phases, and quasi-crystals, have been studied among the rich alloy compositions. Among all the alloying elements, calcium, with its ashes, according to Witte. Its use as a biomaterial dates back to 1878, which is rather remarkable. Then, before to and during World War II, we had a large-scale employment of magnesium alloys in aircraft for solely military objectives [5]. Only the United States' manufacturing capacity would account for nearly a fourth of today's global production capacity of 950,000 tonnes throughout those years. The word "electron" was once associated with magnesium alloys, maybe because of its dazzling white light when burned or as a homage to the old alchemical term electrum. Although the Lydians of the Aegean coast of Anatolia utilised geologically occurring gold and silver alloys for coinage, this

Aluminum, calcium, and magnesium are the most common alloying elements used in magnesium alloys. The addition of aluminum and calcium to magnesium alloys can improve their mechanical properties and corrosion resistance. However, the addition of these elements can also lead to the formation of intermetallic compounds, which can reduce the ductility and formability of the alloy. The addition of magnesium to magnesium alloys can improve their strength and ductility, but it can also lead to the formation of intermetallic compounds, which can reduce the ductility and formability of the alloy. The addition of magnesium to magnesium alloys can improve their strength and ductility, but it can also lead to the formation of intermetallic compounds, which can reduce the ductility and formability of the alloy. The addition of magnesium to magnesium alloys can improve their strength and ductility, but it can also lead to the formation of intermetallic compounds, which can reduce the ductility and formability of the alloy.

Mg was first used for purposes such as ignition and photographic

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