

Differential Pattern of Arsenic Binding by the Cell Wall in Two Arsenite Tolerant *Bacillus* Strains Isolated from Arsenic Contaminated Soil

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Abstract

Arsenite binding was evaluated in two *Bacillus* strains i.e., *B. megaterium* and *B. pumilus*, isolated from arsenic contaminated soil of Unnao district of Uttar Pradesh (India). Initial results showed that more than 90% of arsenite was removed by surface binding by the cell wall component in both the tested species of bacteria. Results on the concentration dependent arsenic binding in bacterial strains exhibited higher effciency of arsenite binding in *B. megaterium* (q_{max} - 1000 mg g⁻¹ protein) than *B. pumilus* (q_{max} - 666.7 mg g⁻¹ protein). The pH optima for arsenic (As) binding in both *B. megaterium* (pH 6.0) and *B. pumilis* (pH 8.0) were found to be different. Results on temperature dependent arsenite binding by *B. megaterium* showed maximum binding at 30°C, while arsenic binding maxima in *B. pumilus* showed a broad temperature range (25°C to 35°C). The kinetic parameters on arsenite binding revealed that both the bacterial strains followed pseudo-second order kinetics. The As adsorption behavior of the bacterial strains was better explained by Langmuir isotherm rather than Freundlich model. Results of FTIR spectra on surface binding involvement of mainly amines, alkenes and C-N functional groups. Whereas FTIR spectrum of *B. pumilus* showed changes in the band region of 3433 cm⁻¹ to 2924 cm⁻¹ indicating the involvement of hydroxyl, alkanes, alkenes, amides and aromatic functional groups in the arsenic binding. A corollary of these results indicated differential binding of arsenite in both the *Bacillus* strains was on account of different arsenite binding ligands on cell surface as evident from

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ReceivedE0.:i 1 E0.54 Tc (M4 Td(E-)Tj150.91 324ngh03 164.57 Uttar Pr)Tj1himra2 324ng

. , i, n min н. i i (27 40' in 🔍 , 80 00') 🏧 . 🧠 i **n**i.A**m.** (.... .http://www. - Ma Ia _ Pe , . . . **),** . . .P. л. . . N Me 15 min. , i . i. -. N i 🖦 r irin 40 "**m** i m .P. ning iN . . . **R** i . re maine in in a line r irir 40 m**N**...

Arsenic biosorption experiment

ir. i 📭 ... i , 10 📭 . **.** i . . . (1500 , ir. 10 min). i, i i i i 🖍 M. B. R. R. i Rik . . . inin , . . , . . i . . . i 2 .m 10 .m . i Air ... i. A I. I. h Ma $(\ldots, (1, \dots, \dots, \dots, \dots, \dots, \dots, \dots, \mathbf{i}) \cdot \mathbf{i} \cdot \mathbf{i}_{1}, \dots, \dots, \mathbf{i} \cdot \mathbf{i}_{n})$ 60 C . **B**e . i. e . A . , , $C \sim_4 i_{4}$, \ldots , i_{4} , \ldots , 18 . i n , ir ... " m i n M.M. R .P. ..**m**,∕i.m_n, ∖. i n Ma in. .,..**N**... . Pe ., . N R.R. in A., i.r.

Determination of adsorption isotherms

Ani m.i n . ii i 🚌 ir. i sis. R. R. 1 n. 💐 🚬 n. . i n i ,....in ...**n**-..... .P.i**m**,,, i in ... i nin . . **.**P. ,... i**n** . i ,... .

Langmuir isotherm

 $\sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{$

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$$\frac{C_e}{q_e} = \frac{1}{q_{max}b} + \frac{C_e}{q_{max}}$$

Freundlich isotherm

$C_{n} = C_{n}^{1/R}$

$$\log q_e = (1/n) \log C_e + \log K_f$$

r Ci , i i ir. 🖄 .r. 1/.r. .. inMa -C,...i.i..........i. . A. i. A. i . . . in₁i i . - i 🗛 -i n A.A R.A. R. i. R. . in i i **i**. . in A. .**P** .P. . .P. . . in i i i.e

Determination of adsorption kinetics

in i, ..., m_{1} , ..., i in ..., i in ..., i in ..., m_{1} , \dots m_{2} , \dots i in ..., i in ..., m_{1} , \dots m_{2} ,

$$\log(q_e - q_t) = \log Q_e - \frac{k_1}{2.303}t$$

$$\frac{t}{q} = \frac{1}{k_2 q_e^2} + \frac{1}{q_e} t$$

 $\sum_{k_1 \in \mathbb{N}} k_2 (\mathbf{p}_k^{-1} \mathbf{p}_1^{-1} \mathbf{p}_1^{-1}) \dots \mathbf{p}_k = \mathbf{p}_k + \mathbf{p}_k^{-1} \mathbf{p}_1^{-1} \mathbf{p}_1^{-1$

Fourier transforms infrared (FTIR) spectroscopy

в. in. J. m, i. i., in, ۴., .1 ni i n J. 2 50 C. A i 12 .1 B (1:100). i 🖻 mi. i .M ۴. mi in. i i P. P. i in. r. r. 400-4000 i in B in, . in P. **⊠_**A). 6700, (i i ni .Pp л. B .m. i in, h -.. P. .M. .Me .Pp

Statistical analysis

A . in i.i. - im r .P. . , . , **i**. , . л. <0.05. ji **(** .**P.** . . i .i. (A i ni , i ... 8. .Be , in ~ . P. P.) in (💐 i 1,7). , i-, . . A P.

Results and Discussion

E ect of pH on surface binding and intracellular uptake



Bacillus ____i . A ni 35 C 30 C in. in. . 31 in r i i i in. ir. . i. . , A.P i i n in ir. A. i. i ir. А 1.

E ect of As concentration

100 i 🖍 (10 -1) i .1 л. in. i. .P. i i 3. C rin ir. in, in, in. in. ir. i. in, 50 A, P. M in. Ŀ. 11. u. .M. R m, i n ir .L

in. 33 . A 32 A ir, ir, i Р., in. Р. Α 34 in, i - i 🖪 ir. ur. (76%) <u>r</u> ir (12%) in. B. megaterium. in in 6 n i i 26 -37 . A 35 į ir, ir, in . 38,39 . i i



Figure 4: Langmuir and Freundlich isotherm for the adsorption of Arsenic by Bacillus megaterium (A) and Bacillus pumilus (B).

Bacterial Strains	Langmuir Constant			Freundlich Constant			
	Max. adsorption capacity 'q _{max} ' (mg g ⁻¹ protein)	Adsorption affnity 'b' (L mg ⁻¹)	Regression Coeffcient 'R ² '	Adsorption coeff cient ${}^{\prime}K_{f}$ (mg g ⁻¹ protein)	Adsorption intensity ' <i>n</i> '	Regression Coeffcient 'R ² '	
B. megaterium	1000.0	0.0013	0.910	-0.499	1.17	0.781	
B. pumilus	666.7	0.0008	0.934	-3.0	1.12	0.838	

Table 1: Langmuir and Freundlich isotherm constants for the adsorption of Arsenic by B. megaterium and B. pumilus.

Bacterial Isolates	Q (Experimental)	Pseudo Second Orde	r Kinetic Constants	Pseudo First Order Kinetic Constants			
		Q₀(Calculated) (µg/mg) mg)	K₂(gmg ^{⁻1} min ^{−1})	R	Q _e (Calculated) (µg/mg)	K (min)	R
B. megaterium	35.62	38.46154	0.006318	0.999	-0.08197	0.006909	0.591
B. pumilus	26.76	28.57143	0.007206	0.999	-0.0846	0.009212	0.702

Table 2:

Biosorption isotherms

A i n ir. .1. 1. ni in. b) i (qir (Kn) i n i n Р., 2) (1 Р. i . Ni ni. m ., , 1/C. (i 1/ir. J . J **.** b 4). - i 🖪 i q_{max} Ŀ. . Р. q_{max} . i в. B. megaterium > B. pumilus in. in. B. megaterium 1). B (n q_{max}, i B. pumius. , i**m** i i . Me .ni C 29 q_{max} . In i , i . i .P. .P. b *b*, i . . i in. q_{max} .P. л. i n in .,.

 K_{F} n i n. \boldsymbol{n} (4) C i . В. ni i in. i i. B .P. .B $(K_{\rm F})$ in. ir. m, B i ni ۴., л. лi 1.6 in, J. ir r r, i in, . 16, , i., .P. 11 M. B. in E. coli n in in . . N m, i i P

Biosorption kinetics

Real Bacillus, in Bacillus, in

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i.∎, 50 . ₽, -1(i....5). . Pe J. **R** inP. (, , -,) *VS* **i** . . .i**n** - i n ii. i**n** ... i 📭 in i. i. -- i . i**r**. Pq. .i ... i**r.**.. ...i in .F. .F., ., . -. / . . . **M**.**P**4 2. - Rilli**n** . in .Pp (²) .P. 2

1604 m⁻¹, ii r A, i ir i 1429 m m 1428 m i i 📭 i 888 "m⁻¹ ir. i Р. л. P. 885 p⁻¹ ., ni (3). - i 🖪 i P.

3426 **m**⁻¹ i**n** *B. pumilus* i .m. 3433 .m.⁻¹ in .,.A, i in in. . Mi P. 2. **m** 3080 3066 m C, m i in. P. 2924 🛌 С-, C, in, į . i.r. 3 2926 m⁻¹ in, i ii./ i P. (2960 P⁻¹ 970 ir. i A. i ni. . . .P. .P. .P. "**m**,⁻¹), ", in, н n in Ŀ, • J. in. л. .P. i i i in. / in m. .P. 1460 . ir. i. , .**i** ., ni ir ir -.M. .P. ~ 1453 📻 i n (3). i ni ir. n

ni m љi m. i in. P. 1.i P. in, I. i. R .M i n 45. С i in, i . С-, į 2.1 J. ir. ir. л, j, i. mi л. n in ni m . Ma i. .i.n m. i in i .**e**. 46-P. A 49. min. nim i mi in. in in i 50,51 . C i , C-**P**• **,** mi .M.) r ir n i ir, ir, ni (.Pe in E. coli 16 .

Conclusion

В ir. , i л. ., **,i** 1 90%) 📭 in in (ni. i . Ma n in. .UMe .Ma N i i ir. im m. -i rin ni ∕ in, 1 I. Bacillus in in i i in, I. I. .1 i i н in in , . . . n in P. 1. . . .

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