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Introduction

Colloids are a class of solutions comprising particles ranging from 1 to 1000 nanometers in diameter. These particles, termed colloidal particles, exhibit unique properties due to their size, shape, and surface characteristics. Unlike true solutions where solute particles are molecularly dispersed, colloidal particles remain dispersed throughout the solvent, forming colloidal dispersions. This phenomenon, known as colloidal stability, is crucial in various fields such as pharmaceuticals, cosmetics, food science, and environmental engineering. The nanoscale dimension of colloids imparts them with distinct optical, electrical, and mechanical properties, making them valuable in nanotechnology applications. Understanding the behavior of colloids at the nanoscale is fundamental for designing advanced materials, drug delivery systems, and functional Nano devices [1-3].

Background of colloids

Colloids have intrigued scientists for centuries due to their unique

(FTIR); and scanning electron microscopy (SEM) were employed to analyze the structural, morphological, and compositional properties of colloidal nanoparticles [5-7].

Result and Discussion

Results

The experimental results revealed the formation of stable colloidal dispersions with well-defined particle sizes and uniform distribution. Dynamic light scattering measurements indicated narrow size distributions, with average particle diameters range from 10 to 500 nanometers, depending on the synthesis method and conditions. Zeta potential measurements demonstrated the electrostatic stability of colloids, with high absolute zeta potentials indicating repulsive forces that prevent particle aggregation. Structural analysis using X-ray diffraction confirmed the crystalline nature of certain colloidal