

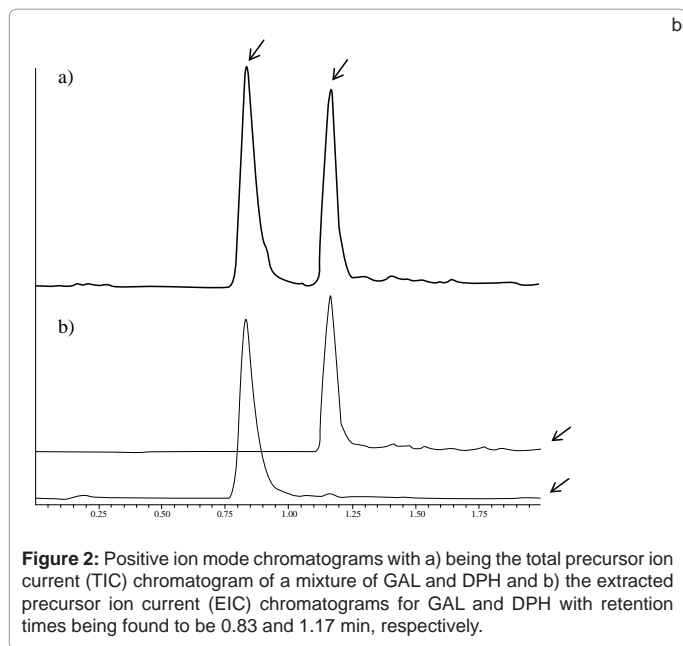
**Keywords:** Guinea Pig; Galanthamine; Liquid chromatography; Mass spectrometry; Chemical warfare agents; Nerve agents

## Introduction

Organophosphorus compounds, since their early introduction in agriculture as a pesticide and later as Weapons of Mass Destruction (WMD) in the form of Chemical Warfare Agents (CWA), are a global concern for their toxicity in humans [1-3]. Organophosphorus G-Series (Sarin (GB), Cyclosarin (GF), Soman (GD), Tabun (GA)), and V-Series (VE, VG, VM, VX) compounds are one classification of CWA by convention, which act to severely disrupt neurological regulation within biological systems through their irreversible inhibition of the enzyme Acetylcholinesterase (AChE) [4-6]. The degree of toxicity (e.g. chronic to acute) or severity of the disruption (e.g. mild to severe) is determined by the specific compound and its concentration.

Citation:

---



**Figure 2:** Positive ion mode chromatograms with a) being the total precursor ion current (TIC) chromatogram of a mixture of GAL and DPH and b) the extracted precursor ion current (EIC) chromatograms for GAL and DPH with retention times being found to be 0.83 and 1.17 min, respectively.

acceptors),  $\pi$ - $\pi$  (interaction with dioxane - proton donors) and  $\pi$ - $\pi$  (interaction with nitromethane - dipole-dipole interactions) that could be fine-tuned for chromatographic separation integrity and extraction recovery selectivity. Consideration was also given in the selection process to organic solvents with lower boiling points in an effort to increase the rate at which the upper organic layer could be evaporated in the sample preparation process. Thereby, minimizing the rate-limiting time step needed for sample preparation. Ultimately, trichloromethane with a  $P'$  value of 4.1 (e.g. 0.1 for non-polar hexane and 9.0 for polar water) and a boiling point of around 61°C was used. Optimal extraction recoveries as determined by comparing the peak area for plasma samples for three replicate inter-batch assays that were spiked before and after extraction were obtained for both GAL (0.025 (8.69 nM), 0.25 (86.9 nM), and 1.00 µg/mL (348 nM)) and DPH (0.25 µg/mL (97.9 nM)) in 10 min with a trichloromethane liquid-liquid extraction.

### Mass spectrometry development

Chromatographic mobile phase composition with a continuous isocratic ratio of (65:35 v/v) mobile phase A (0.05% formic acid, 0.2% glacial acetic acid, 99.75% HPLC grade water) to mobile phase B (0.1% formic acid, 9.9% HPLC grade water, 90% methanol) was found to be optimal for both chromatography and mass spectrometry ionization efficiency after many trials when combined with a Waters Symmetry Shield 3.5 µm packed RP18 guard (3.9 mm x 20 mm) and analytical (4.6 mm x 75 mm) column assembly. From a chromatography perspective the combination of select mobile phase solvent constituents (e.g. methanol organic modifier and formic acid to displace ion pairs) and stationary phase media type (e.g. shielded silanol non-polar RP18 packing) provided an environment in which equilibrium chemistry could take place between the two phases to provide improved Gaussian peak shapes with consistent baseline resolution, as seen in Figure 2, for GAL and DPH in a relatively short amount of time (i.e. 0.83 and 1.17 min, respectively). From a mass spectrometry perspective the use of key mobile phase solvent constituents (e.g. formic and acetic acid) helped to greatly improve the sensitivity of the analysis by increasing the positive ionization efficiency in the source region of the mass spectrometer.

Mass spectrometric positive ion Turbo IonSpray mode quantifier precursor to product ion ( $M+H$ )<sup>+</sup> transitions of 288-to-213 m/z and 256-to-167 m/z were found after a few trials to be the most sensitive and selective for GAL and DPH respectively, in this study. Here, the precursor ions (e.g. 288 and 256, GAL and DPH) were selected in Quadrupole 1 (Q1), fragmented by collisionally induced dissociation (CID) in Quadrupole 2 (Q2), and then product ions (e.g. 213 and 167, GAL and DPH) were selected in Quadrupole 3 (Q3). The quantifier precursor to product ion ( $M+H$ )<sup>+</sup> transition of 288-to-213 m/z for GAL is shown in Figure 3 as a) full scan Q1 showing the GAL precursor 288 m/z ion and b) full scan Q3 showing the GAL product 213 m/z ion. In Figure 4, the quantifier precursor to product ion ( $M+H$ )<sup>+</sup> transition of 256-to-167 m/z for DPH is shown as a) full scan Q1 showing the DPH precursor 256 m/z ion and b) full scan Q3 showing the DPH product 167 m/z ion. The same type of mass spectrometry ion selection was also done for the quantifier precursor to product ion ( $M+H$ )<sup>+</sup> transitions of 288-to-231 m/z and 256-to-152 m/z for GAL and DPH, respectively. It is ensured that if a case did happen to arise in which the quantifier

ion transitions for GAL and DPH were not present the qualitative ion transitions could be used for the quantitation of GAL. The modes of fragmentation of the 288 m/z GAL and 256 m/z DPH precursor ions producing product ions at an m/z of 197, 209, 213, 225, 231, 270 for GAL and 152, 167, 183, 230, for DPH have been established in literature [31,32,34,35].

## Conclusions

A new high throughput sample preparation extraction LC/MS/MS analysis assay has been developed with commercially available materials to enable future researchers the ability to reproducibly and sensitively quantitate galanthamine in guinea pig plasma. The sample preparation employing a 10 min trichloromethane liquid-liquid extraction gave consistent extraction recoveries of GAL and the internal standard DPH from 160  $\mu$ L plasma sample volumes. The LC/MS/MS analysis method operated in the positive ion MRM Turbo Ionspray mode yielding highly selective quantitative precursor to product ion (M+H)<sup>+</sup> transitions for GAL (288-to-213 m/z) and DPH (256-to-167 m/z). Sample run times were on the order of 1.50 min per sample. Most importantly, it is envisioned that this new extraction assay with low sample volume requirements, rapid GAL extraction recovery, and rapid sample runtimes would be beneficial to not only a wide range of researchers researching pharmacokinetics, bioavailability, or bioequivalence studies but of more specific interest to researchers exploring applications of organophosphorus nerve agent poisoning countermeasures in guinea pigs and humans.

## Acknowledgements

The authors would like to acknowledge: Dr. James Bruce's Laboratory from the University of Washington Genome Sciences, Seattle, WA; The University of Washington's Proteomics Resource (UWPR95794); Dr. Jeff Jones's Laboratory from Washington State University, Pullman, WA for open use of a variety of liquid chromatography and mass spectrometry instrumentation; and the United States Army Medical Research Institute of Chemical Defense, Aberdeen Proving Grounds, MD, for their continued scientific correspondence and support.

## References

1. Ohbu S, Yamashina A, Takasu N, Yamaguchi T, Murai T, et al. (1997) Sarin poisoning on Tokyo subway. *South Med J* 90: 587-593.
2. Romano JA Jr, King JM (2001) Psychological casualties resulting from chemical and biological weapons. *Mil Med* 166: 21-22.
3. Sogorb MA, Vilanova E, Carrera V (2004) Future applications of phosphotriesterases in the prophylaxis and treatment of organophosphorus insecticide and nerve agent poisonings. *Toxicol Lett* 151: 219-233.

4. GROB D (1956) The manifestations and treatment of poisoning due to nerve gas and other organic phosphate anticholinesterase compounds. *AMA Arch Intern Med* 98: 221-239.
5. Compton JAF (1987) *Military Chemical and Biological Agents: Chemical and Toxicological Properties*. The Telford Press, Caldwell, New Jersey.
6. Ballantyne B, Marrs TC (1992) *Pharmacology and toxicology of organophosphates*. Butterworth-Heinemann Oxford 3: 35-39.
7. Williams PT, Hilmas CJ (2010) Cholinergic Effects On Ocular Flutter In Guinea Pigs Following Nerve Agent Exposure: A Review. *J Med CBR Def* 8: 1-13.
8. Shih TM, Duniho SM, McDonough JH (2003) Control of nerve agent-induced seizures is critical for neuroprotection and survival. *Toxicol Appl Pharmacol* 188: 69-80.
9. Bajgar J (2004) Organophosphates/nerve agent poisoning: mechanism of action, diagnosis, prophylaxis, and treatment. *Adv Clin Chem* 38: 151-216.
10. Newmark J (2007) Nerve agents. *Neurologist* 13: 20-32.
11. Berman HA, Decker MM (1986) Kinetic, equilibrium, and spectroscopic studies on dealkylation ("aging") of alkyl organophosphonyl acetylcholinesterase. Electrostatic control of enzyme topography. *J Biol Chem* 261: 10646-10652.
12. Jones DE, Carter WH Jr, Carchman RA (1985) Assessing pyridostigmine efficacy by response surface modeling. *Fundam Appl Toxicol* 5: S242-S251.
13. Kassa J (2002) Review of oximes in the antidotal treatment of poisoning by organophosphorus nerve agents. 40: 803-816.
14. Wetherell J, Hall T, Passingham S (2002) Physostigmine and hyoscine improves protection against the lethal and incapacitating effects of nerve agent poisoning in the guinea-pig. *Neurotoxicology* 23: 341-349.
15. Deshpande SS, Viana GB, Kauffman FC, Rickett DL, Albuquerque EX (1986) Effectiveness of physostigmine as a pretreatment drug for protection of rats from organophosphate poisoning. *Fundam Appl Toxicol* 6: 566-577.
16. Dawson RM (1994) Review of oximes available for treatment of nerve agent poisoning. *J Appl Toxicol* 14: 317-331.
- 17.

involvement of nicotinic acetylcholine receptors. *Neuropharmacology* 46: 103-114.

20. Kihara T, Sawada H, Nakamizo T, Kanki R, Yamashita H, et al. (2004) Galantamine modulates nicotinic receptor and blocks Abeta-enhanced glutamate toxicity. *Biochem Biophys Res Commun* 325: 976-982.

21. Albuquerque EX, Pereira EF, Alkondon M, Rogers SW (2009) Mammalian nicotinic acetylcholine receptors: from structure to function. *Physiol Rev* 89: 73-120.

22.